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# Green Technologies for the Conversion of Crop Residues to Biochemicals

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Canada



# Our Location



Pacific Agri-Food Research Centre, Summerland, British Columbia



# Our Research

- **Fractionation of Crop Residues/Straw**  
-wheat, barley, triticale, oat, and flax straw and shives
- **Using Green Solvents**

# Composition of triticale straw and flax straw

(Results expressed as a percentage of the native, oven-dried basis).

Component	Triticale straw (Pronghorn)		Flax straw (Tieszen07Whole)	
	Native	H <sub>2</sub> O-EtOH Extracted	Native	H <sub>2</sub> O-EtOH Extracted
Total glycans	63.36 (0.01) <sup>a</sup>	63.88 (0.36) <sup>a</sup>	54.99 (0.51) <sup>b</sup>	54.37 (0.28) <sup>b</sup>
Glucan	39.05 (0.13) <sup>b</sup>	40.08 (0.30) <sup>a</sup>	33.35 (0.32) <sup>c</sup>	32.94 (0.06) <sup>c</sup>
Xylan	21.46 (0.04) <sup>a</sup>	21.37 (0.10) <sup>a</sup>	17.73 (0.20) <sup>b</sup>	17.51 (0.04) <sup>b</sup>
Galactan	0.81 (0.06) <sup>c</sup>	0.69 (0.01) <sup>d</sup>	1.52 (0.03) <sup>a</sup>	1.39 (0.05) <sup>b</sup>
Arabinan	2.03 (0.03) <sup>a</sup>	1.74 (0.04) <sup>b</sup>	0.47 (0.02) <sup>c</sup>	0.41 (0.03) <sup>c</sup>
Mannan	ND	ND	1.91 (0.03) <sup>a</sup>	2.13 (0.26) <sup>a</sup>
Total Lignin	21.01 (0.07) <sup>c</sup>	17.08 (0.08) <sup>d</sup>	28.23 (0.08) <sup>a</sup>	25.39 (0.06) <sup>b</sup>
Acid Insoluble Lignin	19.57 (0.07) <sup>c</sup>	15.95 (0.07) <sup>d</sup>	27.28 (0.08) <sup>a</sup>	24.63 (0.05) <sup>b</sup>
Acid Soluble Lignin	1.41 (0.00) <sup>a</sup>	1.14 (0.01) <sup>b</sup>	0.94 (0.01) <sup>c</sup>	0.76 (0.00) <sup>d</sup>
Ash	3.92 (0.03) <sup>a</sup>	0.74 (0.04) <sup>d</sup>	2.02 (0.10) <sup>b</sup>	1.18 (0.09) <sup>c</sup>
Extractives	NA	9.97 (1.01) <sup>a</sup>	NA	5.56 (0.82) <sup>b</sup>

Statistical difference;  $p < 0.05$  determined by Turkey test and Turkey-Kramer test ( $n=3$ ); except total glycans, glucan, xylan, galactan, arabinan, mannan, total lignin, acid soluble lignin, and total of Triticale native ( $n=2$ ).



# Green(er) Solvents of Interest

**Supercritical CO<sub>2</sub>**

**Water**

**Ionic liquids**

# Green Processes

Being Investigated/Optimized

- **Pressurized low polarity water (PLPW) extraction**
- **Supercritical CO<sub>2</sub>**
- **Ionic liquids**
- **Membrane-based separation**

# Green Processes Being Investigation

A decorative header image featuring a close-up of golden wheat stalks and a portion of a bowl with a red and orange rim, set against a warm, orange-toned background.

## **Pressurized Low Polarity Water (PLPW) Extraction Technology**

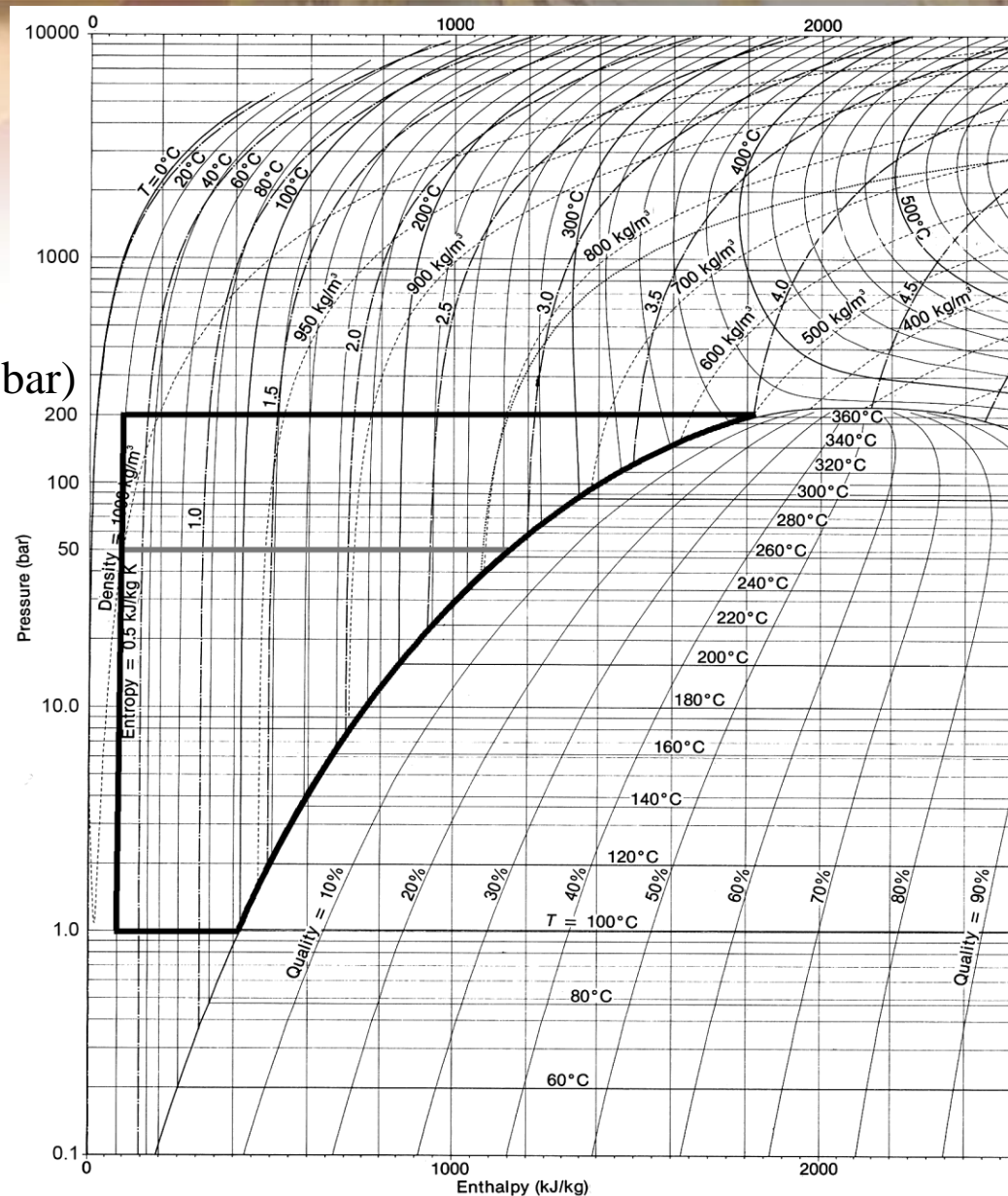
# Pressurized Low Polarity Water Extraction Technology

## What is it ?

- **Pressurized low polarity water (PLPW) is a promising extraction and fractionation technique that uses low-to-moderate pressures to maintain water in the liquid state at temperatures above its normal boiling point.**
- **The higher temperatures decrease the polarity and viscosity and improve the mass transfer properties of water**

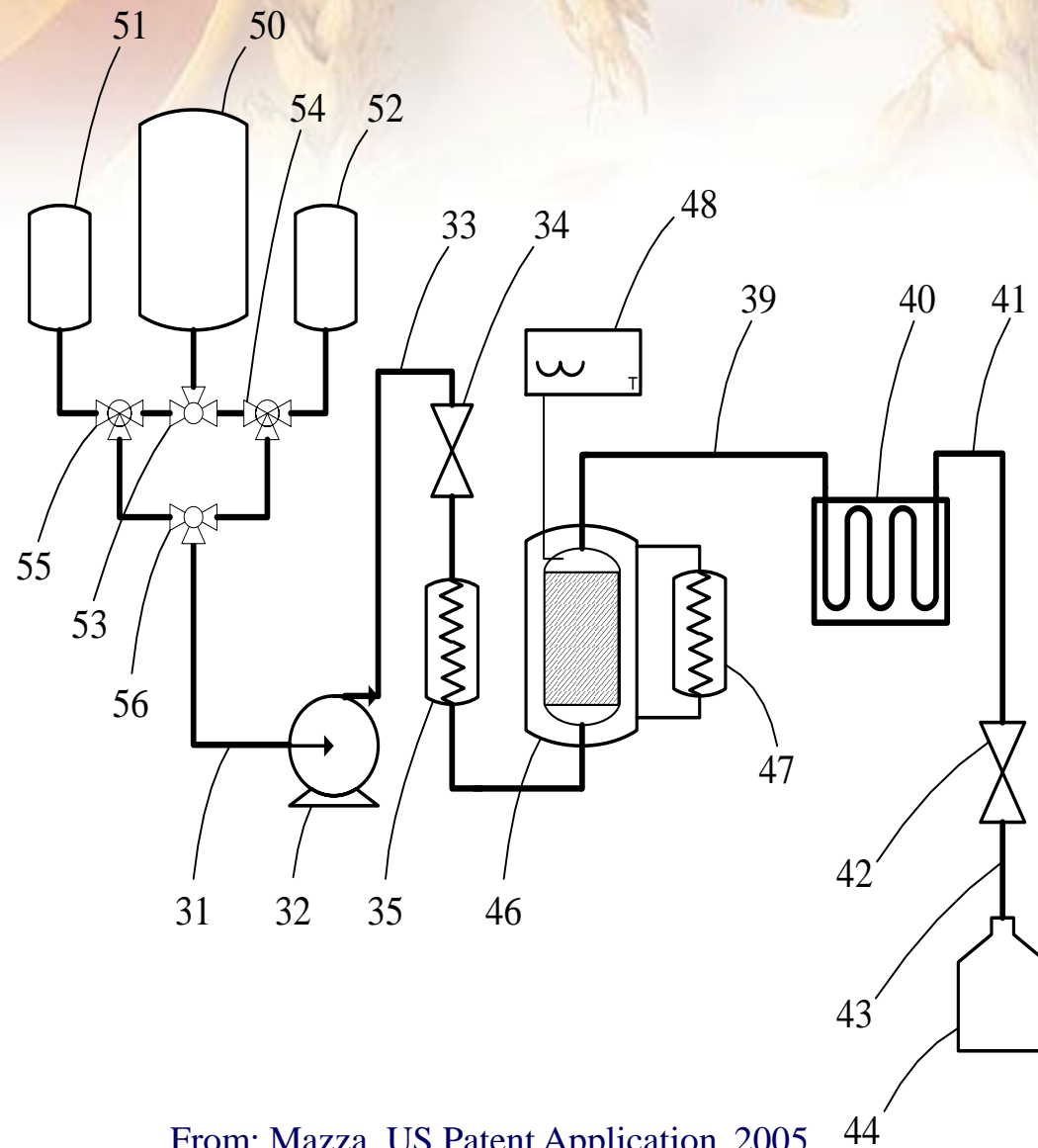


# Pressure-enthalpy chart of water



5.17 MPa (750 psi or 52 bar)

# Diagram of Pressurized Low Polarity Water Extractor



# Pressurized Low Polarity Water Extractor



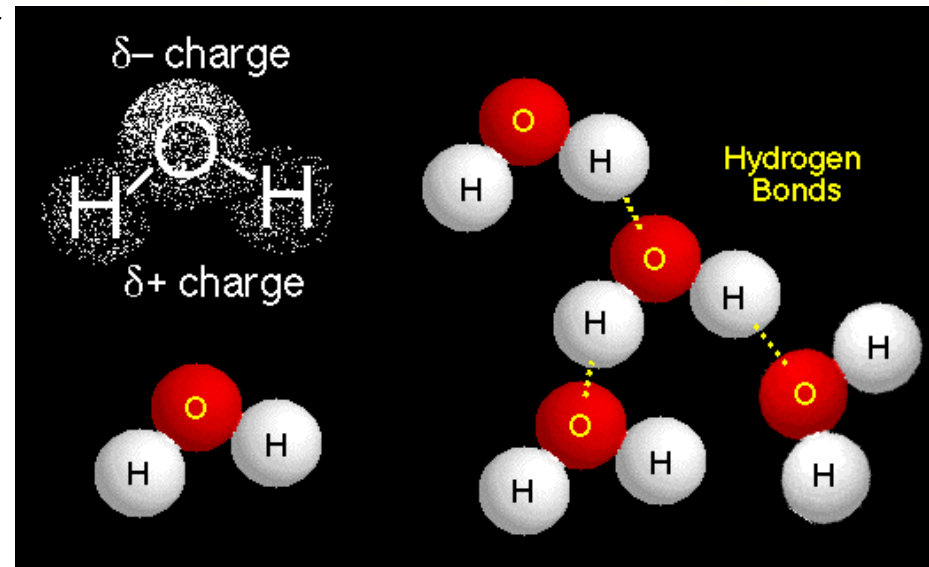
From: Mazza, US Patent Application, 2005



# Changes in the Properties of PLPW

- **Temperature**  $\uparrow$
- **Breakdown of H-bonded water structure**
- **Polarity/dielectric constant**  $\downarrow$
- **Viscosity**  $\downarrow$
- **Surface tension**  $\downarrow$
- **Solubility of organic compounds**  $\uparrow$

The solubility  $\uparrow$  because of the  $\downarrow$  polarity and the  $\downarrow$  enthalpy of solution with the  $\uparrow$  in temp



# Phytochemicals of Interest

- **Lignans from flaxseed and flaxseed meal**
- **Saponins from cow cockle seed**
- **Flavonoids from crop residues/culls**
- **Resveratrol and viniferin grape vines**
- **Hemicelluloses/arabinoxylan from straw**
- **Lignins from straw**
- **Ferulic acid, vanillin and other phenolics from straw**

# Effects of temperature and pH on PLPW extraction of hemicellulose from ground dry flax shives

Design-Expert® Software

Hemicellulose (g/kg DFS)

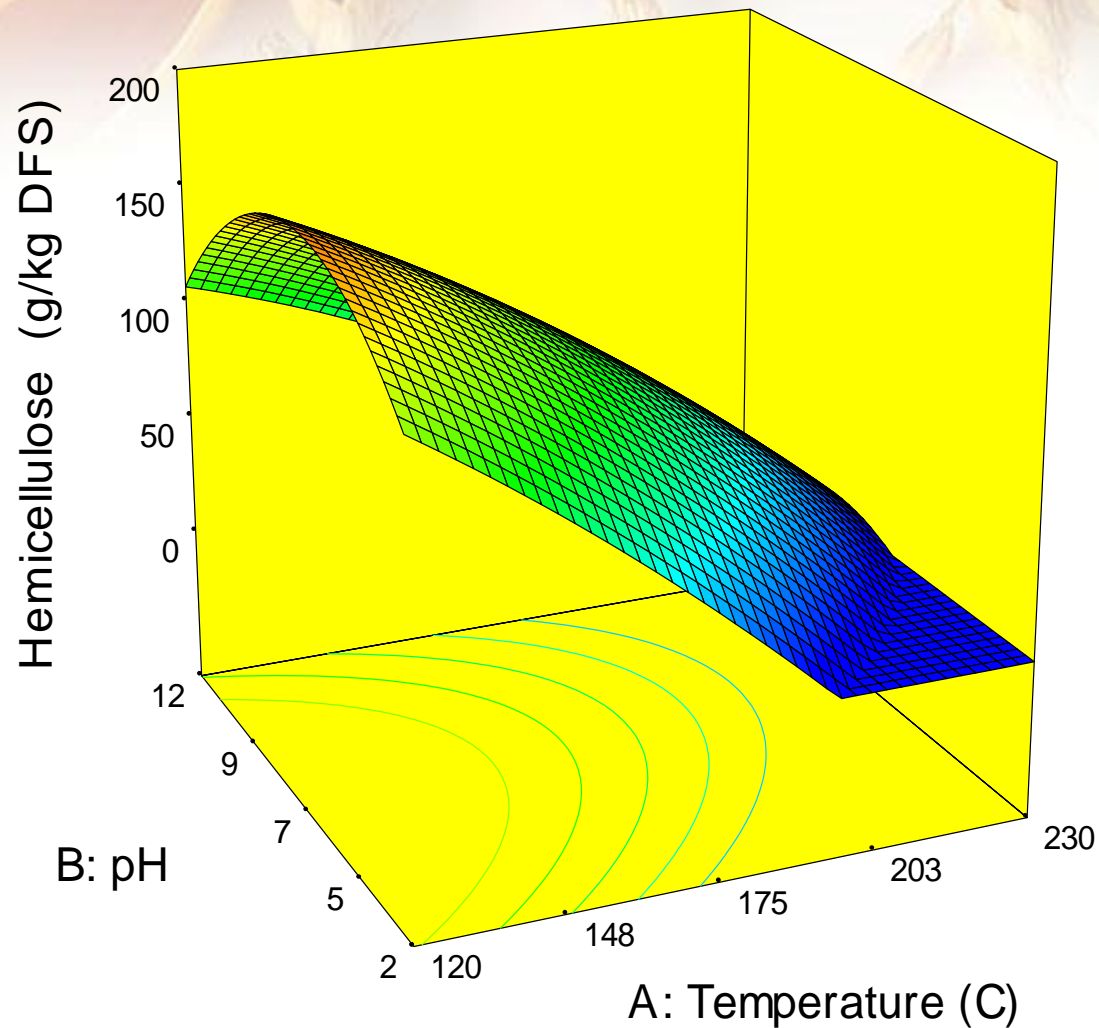


X1 = A: Temperature (C)

X2 = B: pH

Actual Factor

C: Flow rate (ml/min) = 2.0



Hemicellulose in PLPW flax shive residues



# Effects of pH and flow rate on PLPW extraction of hemicellulose from ground dry flax shives.

Design-Expert® Software

Hemicellulose (g/kg DFS)

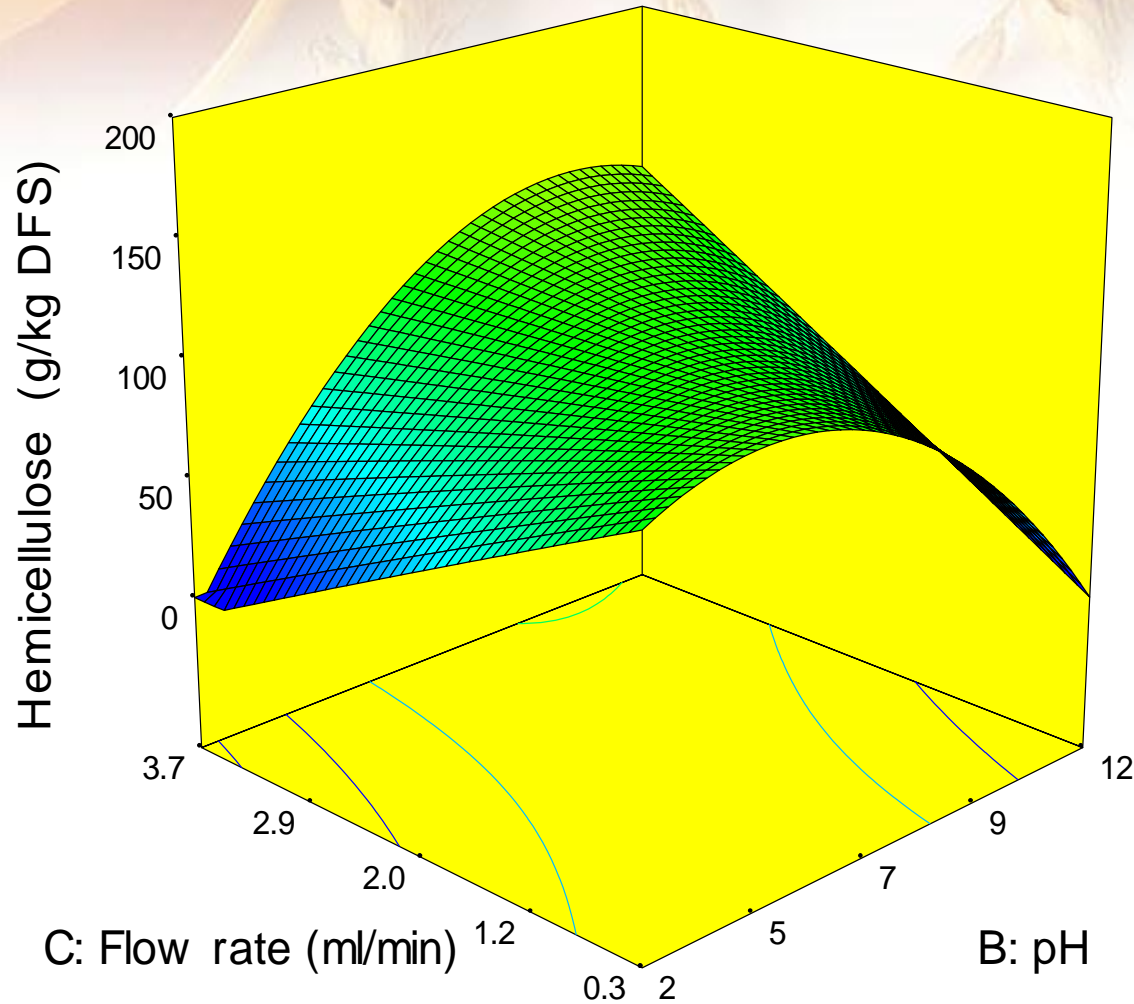
198.291  
11.876

X1 = B: pH

X2 = C: Flow rate (ml/min)

Actual Factor

A: Temperature (C) = 170



# Effects of temperature and pH on PLPW extraction of vanillin from ground dry flax shives

Design-Expert® Software

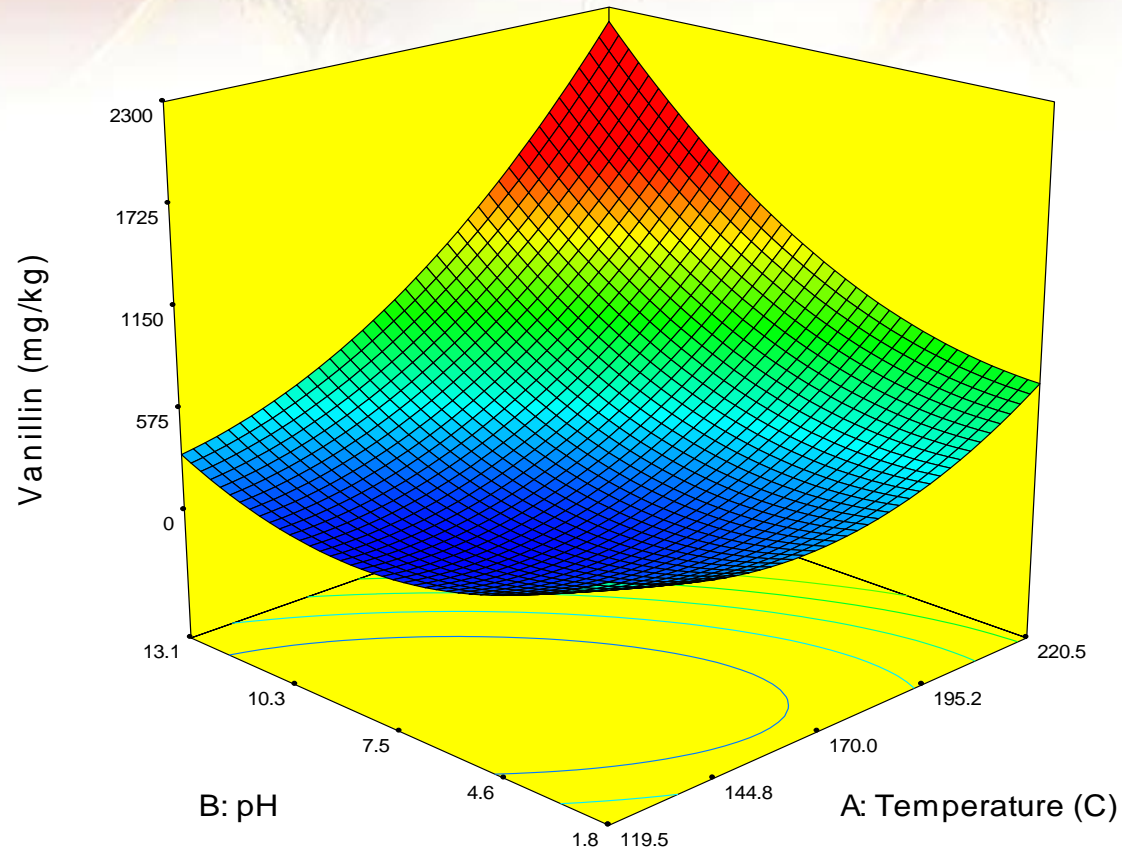
Vanillin (mg/kg)

1450.93  
52.1311

X1 = A: Temperature (C)

X2 = B: pH

Actual Factor  
C: Flow rate (ml/min) = 2.0



# Green Processes Being Investigation

## **Supercritical CO<sub>2</sub>**

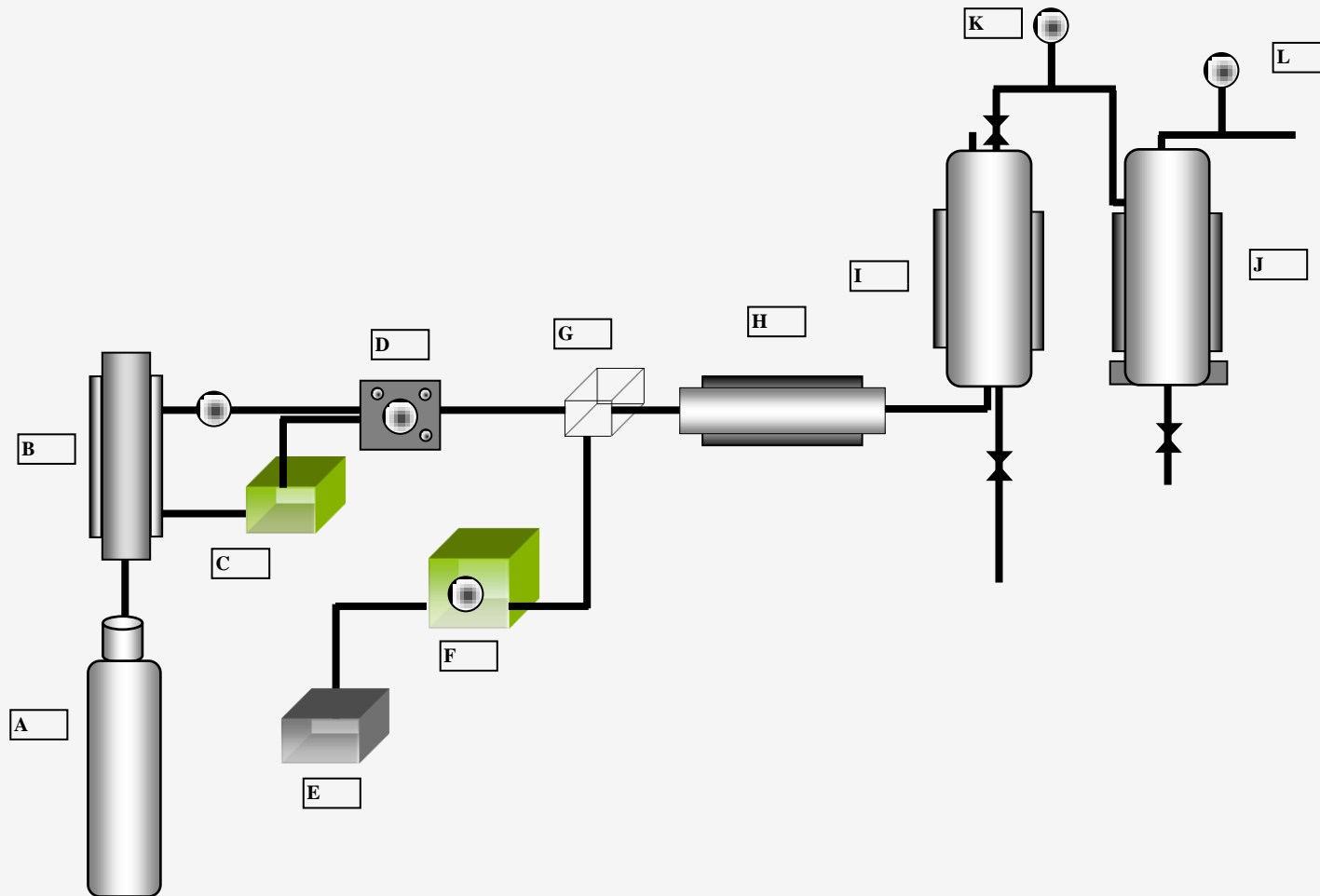


# Supercritical CO<sub>2</sub>

- **Moderate critical pressure (74 bar) and temperature (31° C)**
  - **Non toxic**
  - **Inert**
  - **Non-flammable**
- **Easy removal: simple depressurization**
- **Its physical properties (polarity, etc.) can be tuned through pressure and/or temperature variations**
- **Replacement for organic solvents in extraction processes**

# Extraction

## Supercritical CO<sub>2</sub>



A: CO<sub>2</sub> cylinder, B: heat exchanger, C: cooling bath, D: CO<sub>2</sub> pump, E: solvent reservoir, F: co-solvent pump, G: mixer, H: heating exchanger, I: extraction vessel, K: automated back pressure, J: cyclone separator, L: gauge, M: heating bath

# SC-CO<sub>2</sub> Extraction/Fractionation

## Waxes

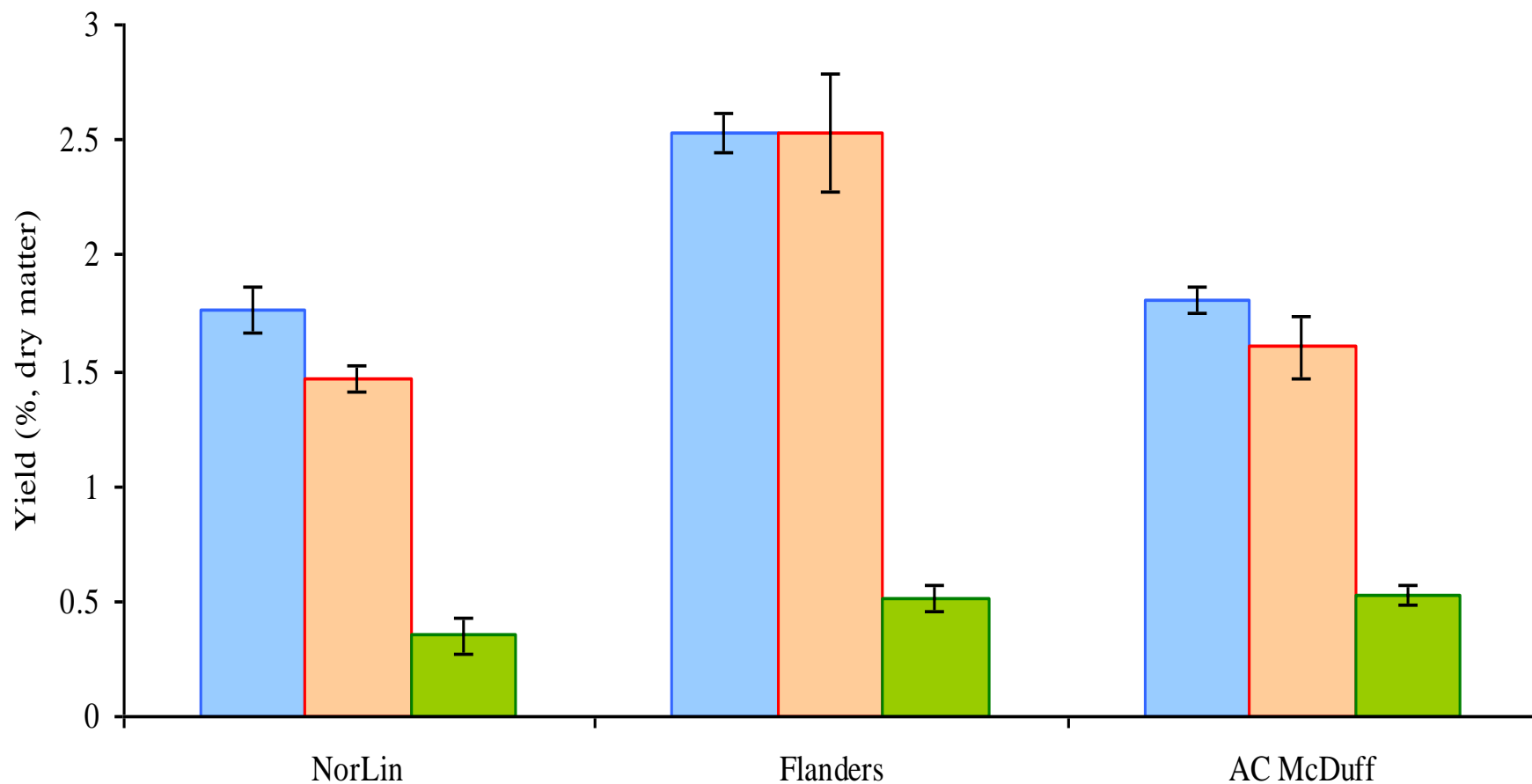
**Polycosanols**—Fatty alcohols

**Phytosterols**

**Alkylresorcinols**--Phenolic lipids

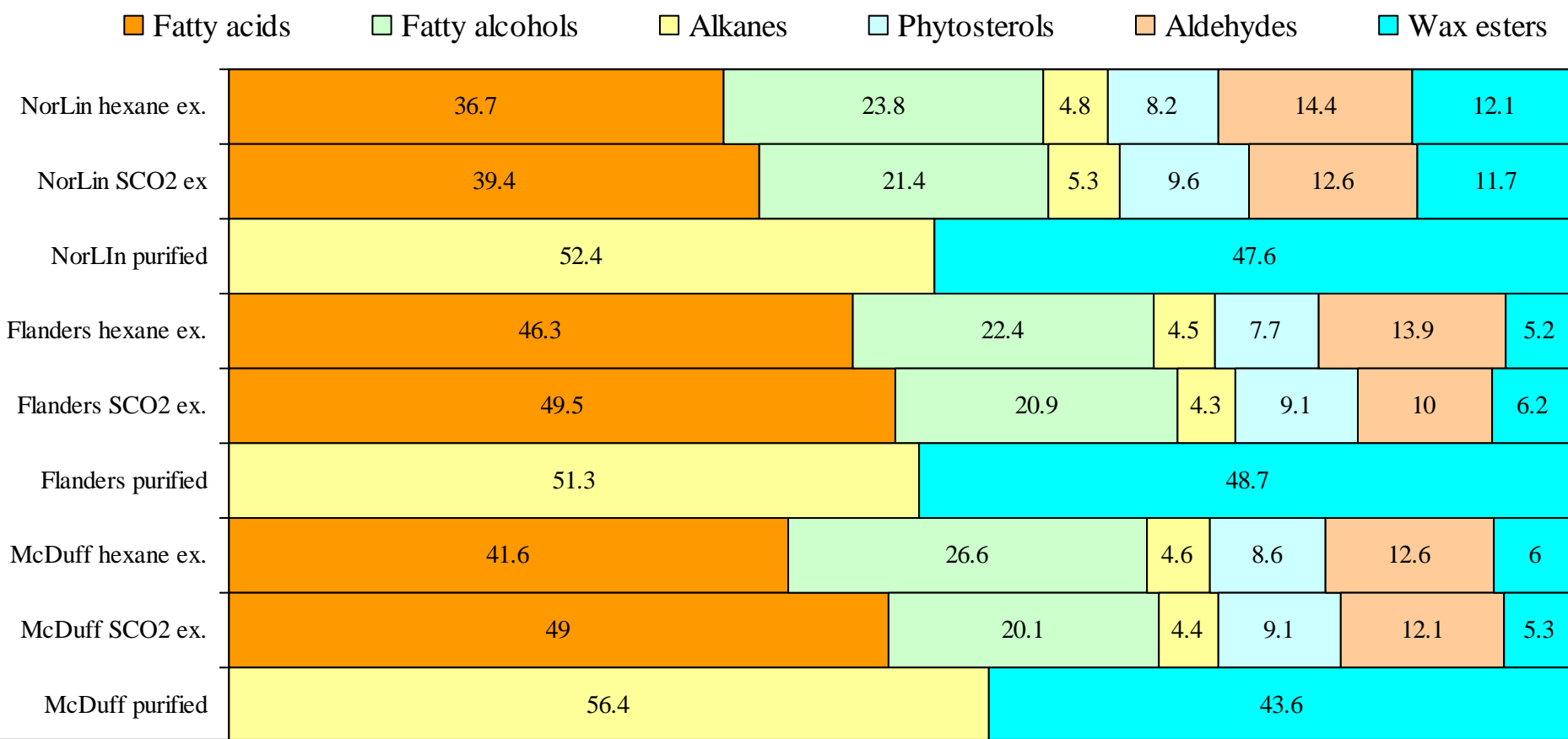


# Yield of wax from flax straw after SC-CO<sub>2</sub> and hexane extraction



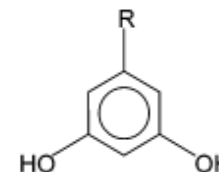
hexane extracted; SC-CO<sub>2</sub> extracted; purified wax samples

# Major constituents in SC-CO<sub>2</sub> and hexane extracts and in purified waxes from flax straw

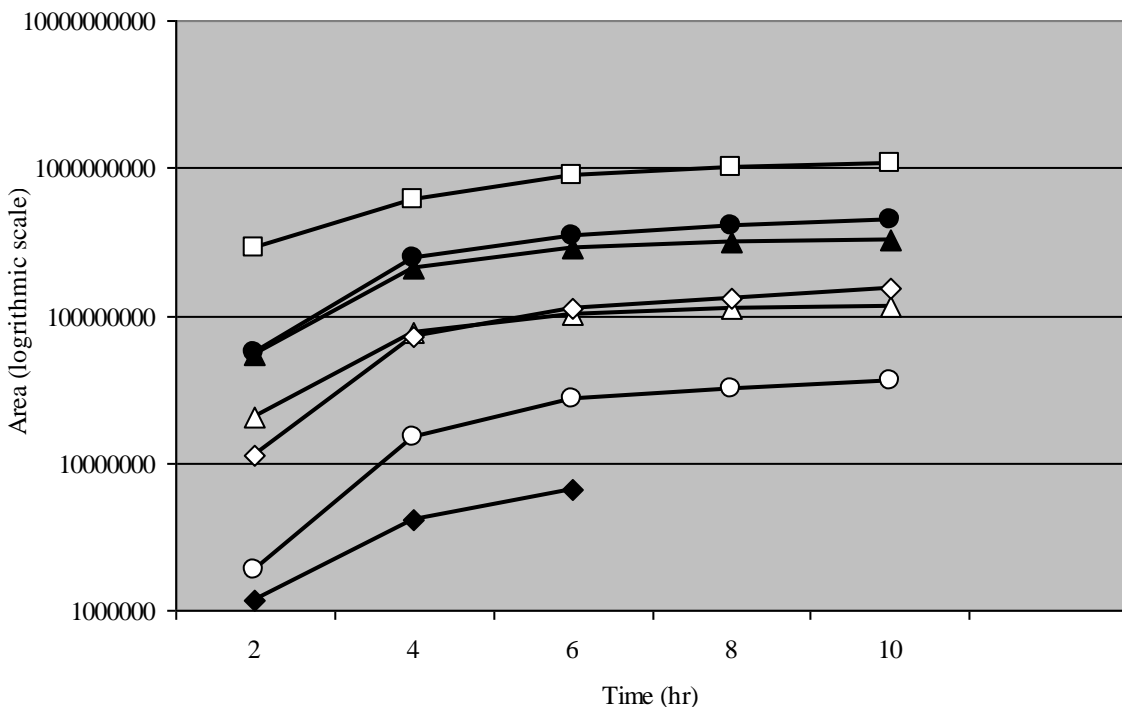


# Alkylresorcinols in triticale bran

Phenolic lipids reported to have anticancer, antimicrobial and antioxidant activity



◆ 15:0    ▲ 17:0    ▲ 19:00    ● 21:0    ◇ 23:0    ○ 25:0    □ total AR yield



R	Names	[CAS No.]
C <sub>15</sub> H <sub>31</sub>	<b>5-n-pentadecylresorcinol</b> , 5-pentadecyl-1,3-benzendiol, cardol (trivial) (C15:0)	[3158-56-3]
C <sub>17</sub> H <sub>35</sub>	<b>5-n-heptadecylresorcinol</b> 5-heptadecyl-1,3-benzendiol (C17:0)	[41442-57-3]
C <sub>19</sub> H <sub>39</sub>	<b>5-n-nonadecylresorcinol</b> 5-nonadecyl-1,3-benzendiol (C19:0)	[35176-46-6]
C <sub>21</sub> H <sub>43</sub>	<b>5-n-heneicosylresorcinol</b> 5-heneicosyl-1,3-benzendiol (C21:0)	[70110-69-7]
C <sub>23</sub> H <sub>47</sub>	<b>5-n-tricosylresorcinol</b> 5-tricosyl-1,3-benzendiol (C23:0)	[70110-60-0]
C <sub>25</sub> H <sub>51</sub>	<b>5-n-pentacosylresorcinol</b> 5-pentacosyl-1,3-benzendiol (C25:0)	[70110-61-1]

# Green Processes Being Investigation

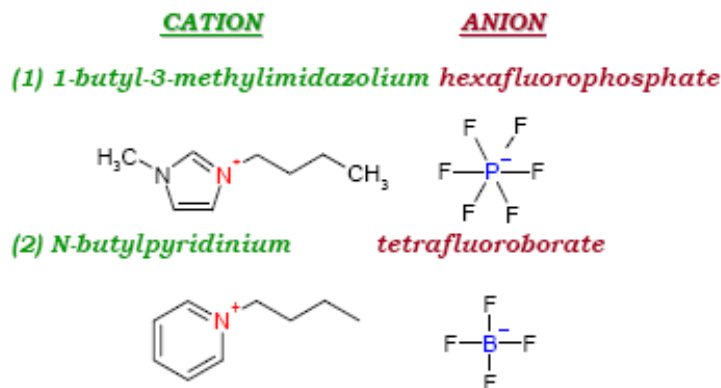
A decorative header image featuring a close-up of golden wheat stalks and a portion of a bowl with a reddish-orange rim, set against a warm, orange-toned background.

## **Ionic liquids**



# Green Solvents - Ionic Liquids (ILs)

- ILs are salts that melt below 100° C, composed wholly of ions.
- CATIONS such as substituted imidazoliums, substituted pyridiniums or others
- ANIONS such as borates, phosphates and halides and others
- Examples and structures



# Green Solvents - Ionic Liquids (ILs)

## Advantage

- No vapor pressure, thermally robust, liquid from  $-96^{\circ}\text{C}$  up to  $300^{\circ}\text{C}$ 
  - Polarity and/or hydrophobicity can be tuned
- “Designer solvents” (1 million ionic liquids vs. 600 organic solvents)

## Disadvantage

- High price
- Toxicity? Biodegradability?
- Recovery of the product and recycling of the catalyst?

# Green Processes Being Investigation

Ionic liquid extraction of lignin from wheat, tritcale and flax straw

Ionic liquid pretreatment of straw

Enzymatic activity in N,N-dimethylethanolammonium alkylcarboxylate ionic liquids

# Green Processes Being Investigation

In this study, we are examining the dissolution of straw lignin in several different ionic liquids, including:

## Cations:

1-ethyl-3-methylimidazolium acetate ([emim]Ac)

1-methyl-3-benzylimidazolium chloride ([bmim]Cl)

## Anions:

dimethylethylammonium formate (DMEAF)

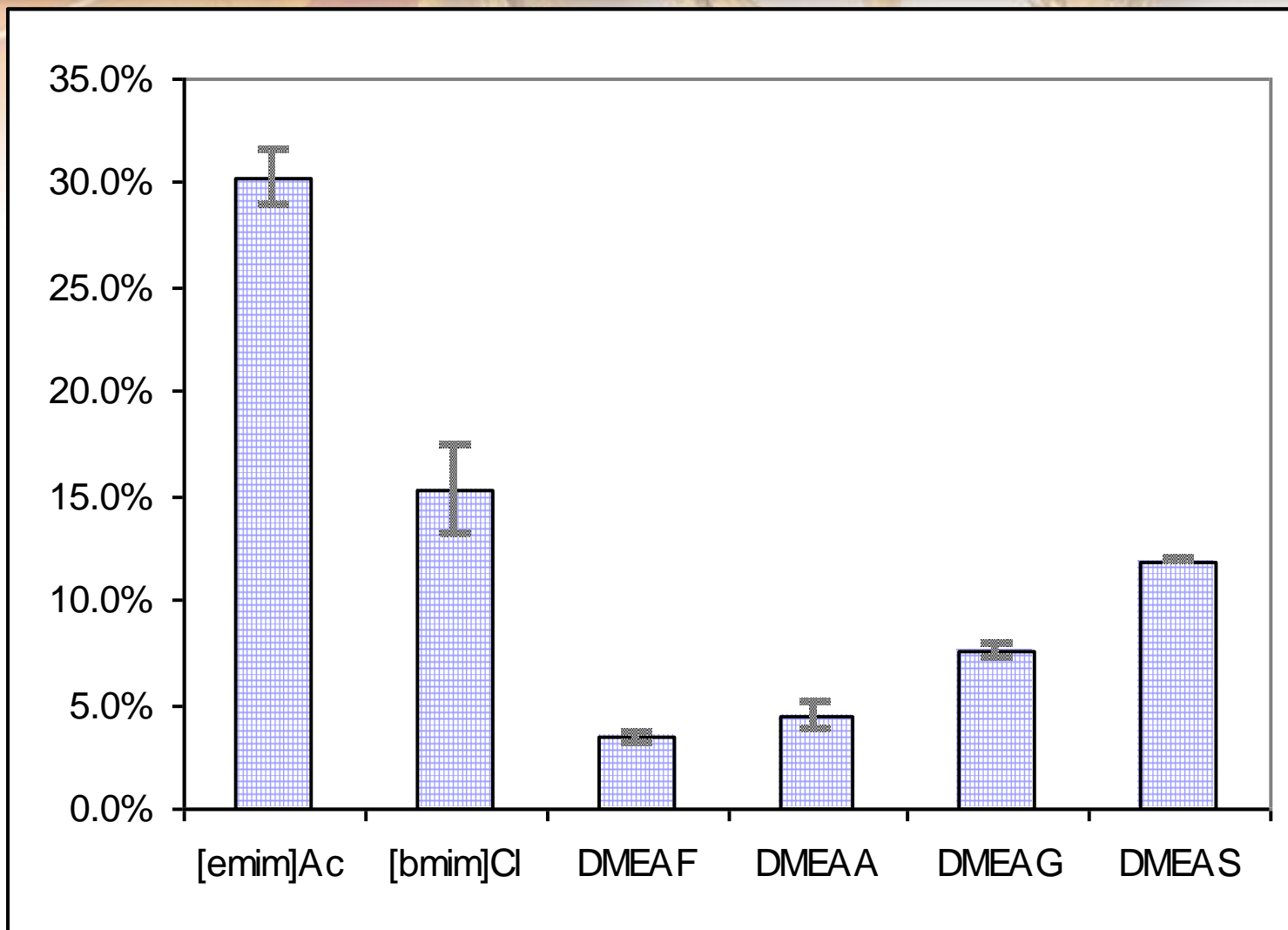
dimethylethylammonium acetate (DMEAA)

dimethylethylammonium glycolate (DMEAG)

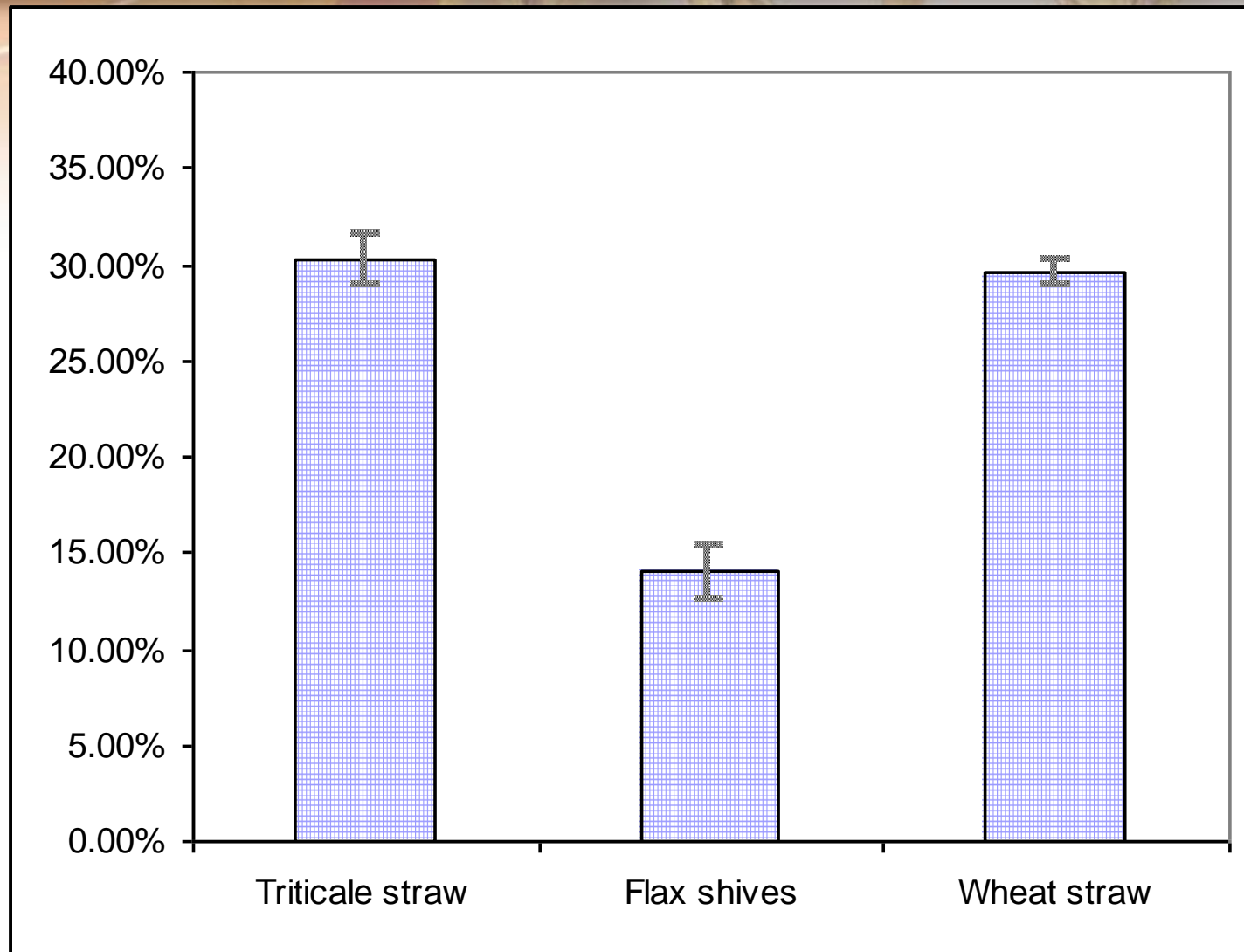
dimethylethylammonium succinate (DMEAS)



# Extraction of Lignin From Tritcale Straw with Six Ionic Liquids at 90°C

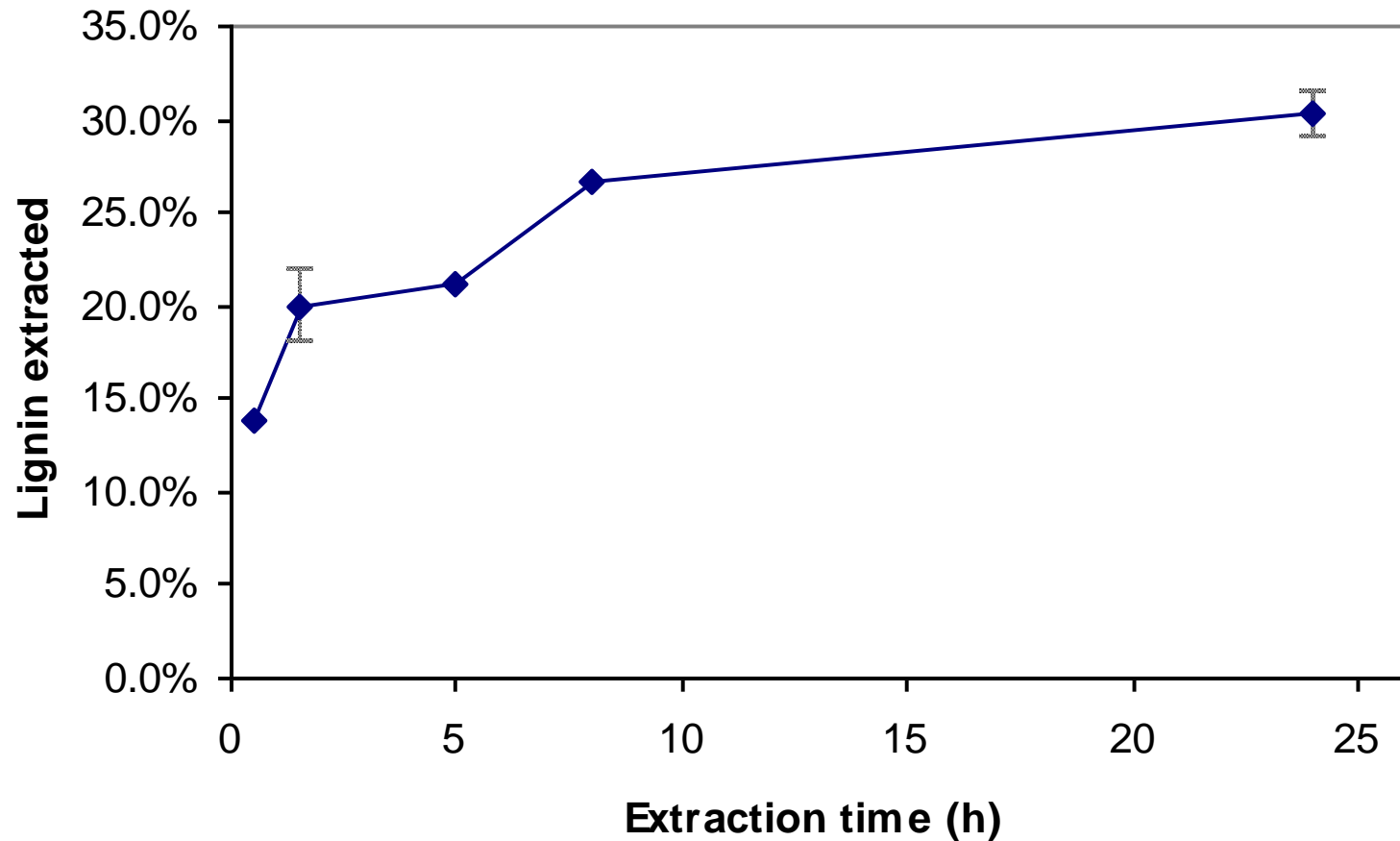


# Ionic Liquid Extraction of Lignin From Wheat & Tritcale Straw & Flax Shives

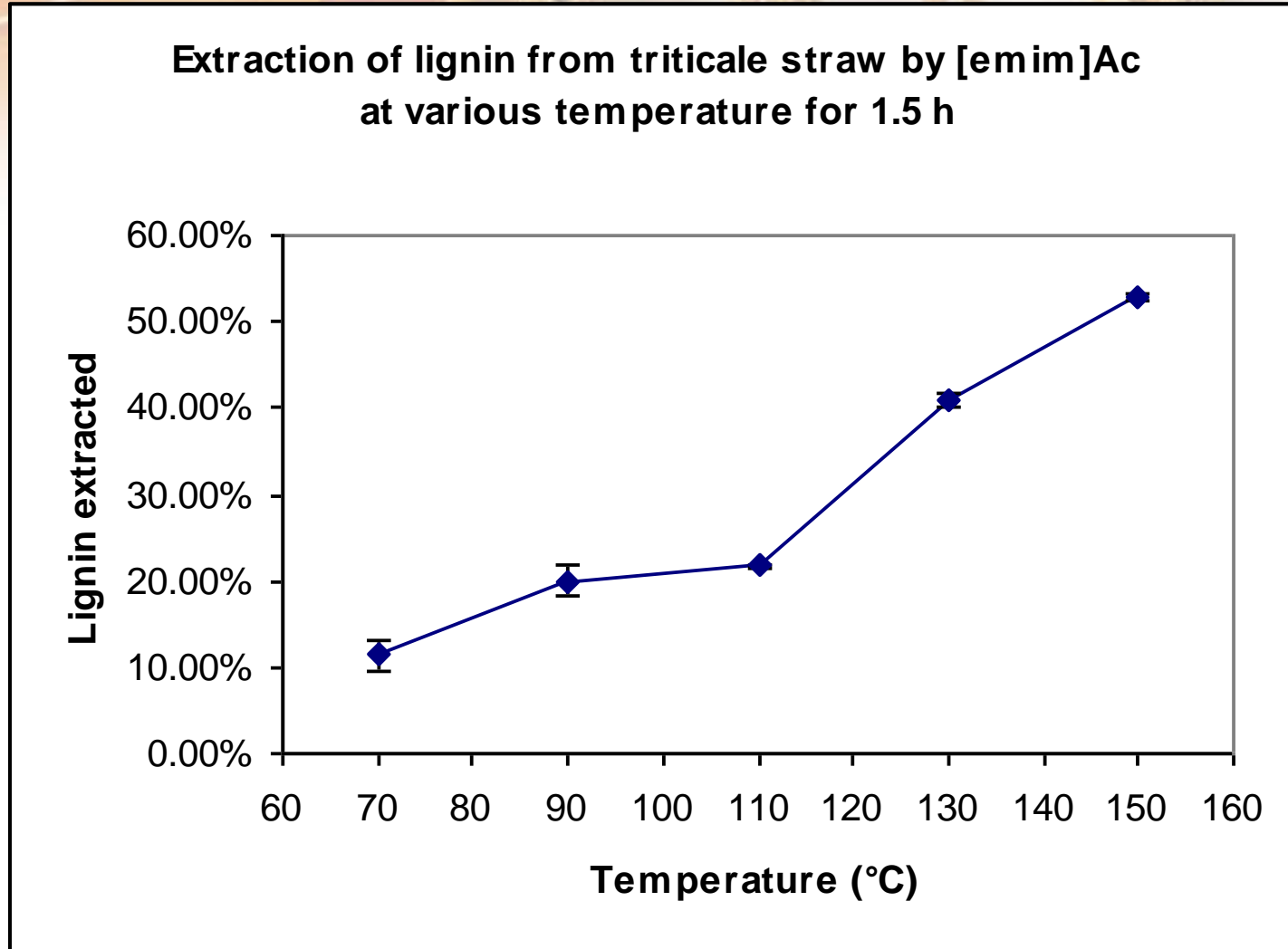


# Ionic Liquid Extraction of Lignin From Wheat & Tritcale Straw & Flax Shives

**Extraction of lignin from tritcale straw  
by [emim]Ac at 90 °C for various time**



# Ionic Liquid Extraction of Lignin From Wheat & Tritcale Straw & Flax Shives





# Green Solvents - Ionic Liquids (ILs)

Rayne, S. and **G. Mazza**. 2008. Trichoderma reesei derived cellulase activity in three N,N-dimethylethanolammonium alkylcarboxylate ionic liquids. Nature Precedings. <http://precedings.nature.com/documents/632/version/1>.



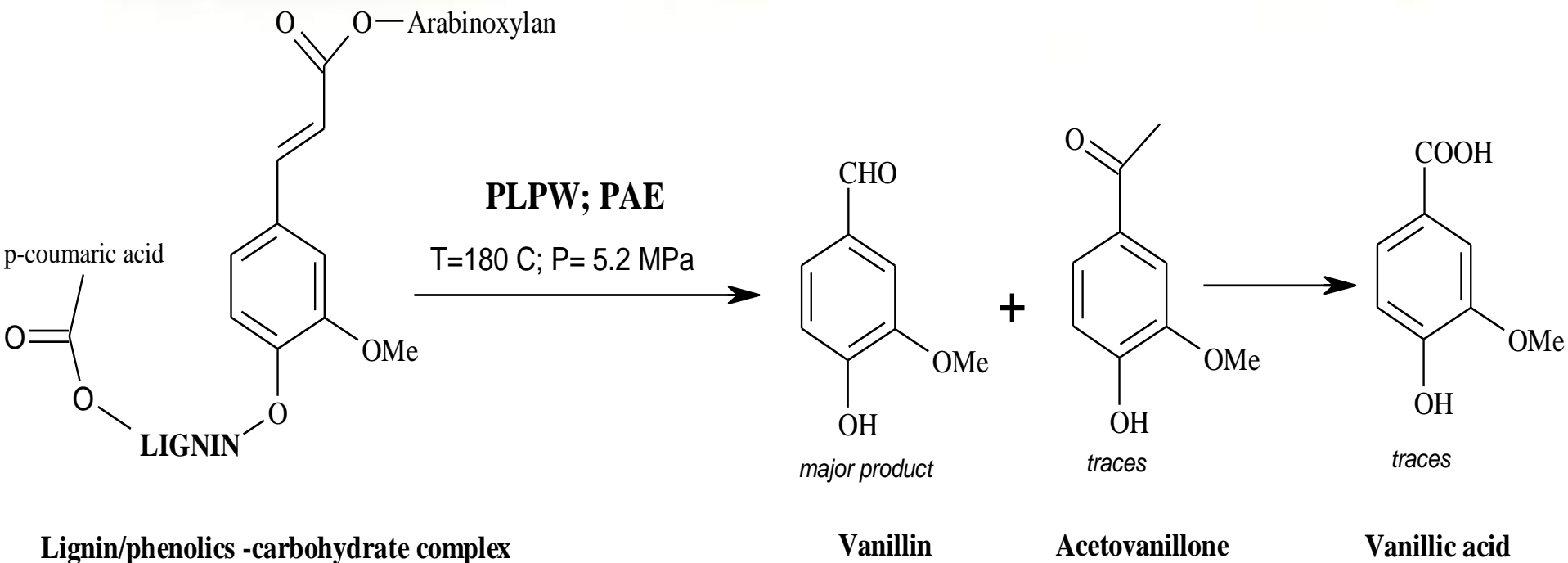
# Conversion Processes

**Thermo-chemical conversion**

**Bioconversion**

# Thermo-Chemical Conversion Processes

Extraction and conversion of ferulic acid from straw and bran by PLPW



# Thermo-Chemical Conversion Processes: Vanillin

Vanillin is an important compound, widely used by the food industry for its aroma. Total consumption is estimated to be 12,000 metric tonnes/year.

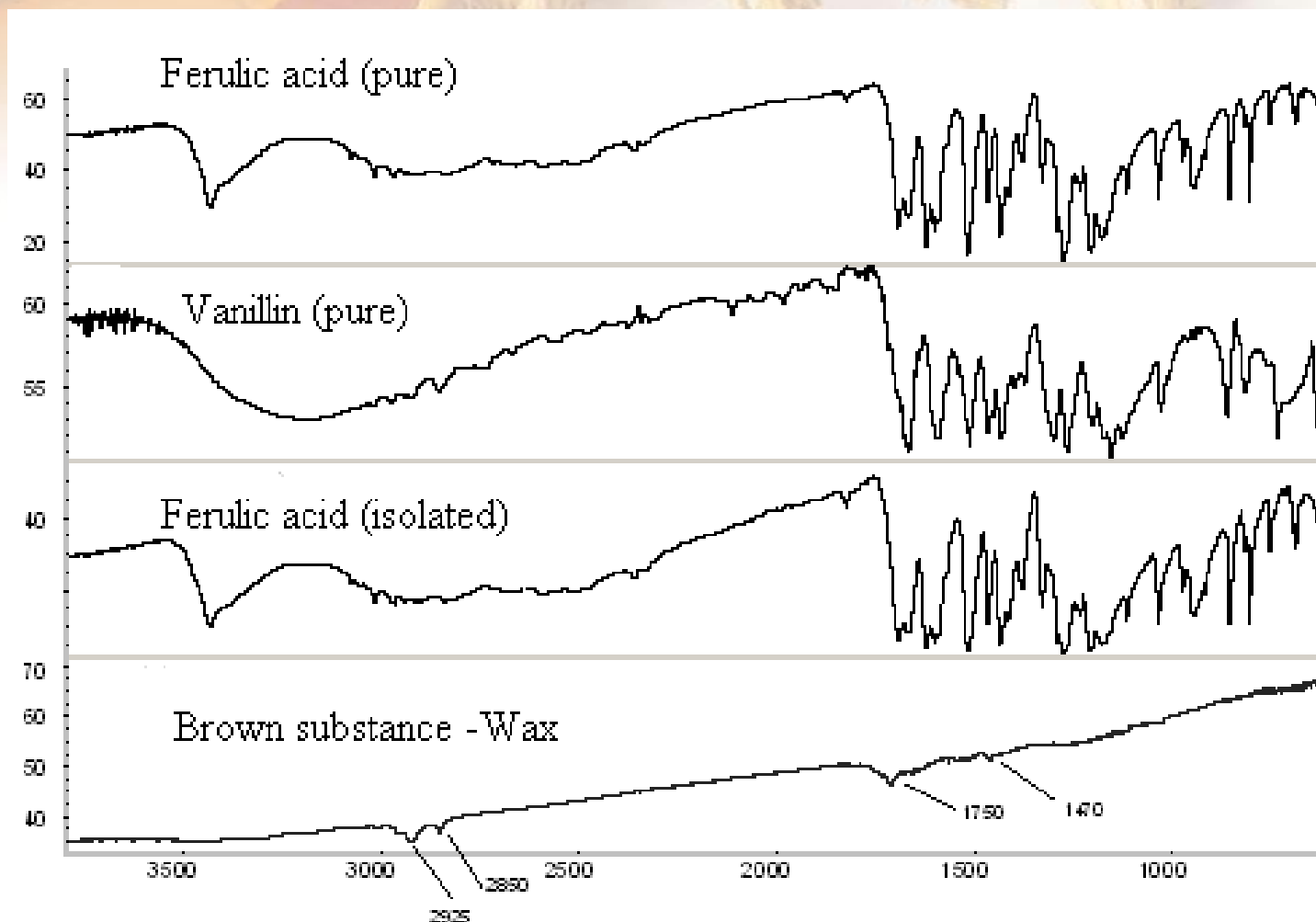
Also used as raw material by the pharmaceutical industry for the synthesis of drugs such as L-dopa used for the treatment of Parkinson's Disease.

Commercial production of vanillin from lignosulfonates, formed during pulping processes, is no longer allowed due to environmental concerns.

The cost for “nature identical” vanillin from plant sources is ~\$1000/kg.



# FT-IR spectra of ferulic acid and wax fractions



# Analytical Methods

**RP-HPLC**

**Capillary gas chromatography (GC)**

**Fourier transform infrared (*FT-IR*) spectrometer**

**Capillary gas chromatography-mass spectrometry (GC-MS)**

**Differential scanning calorimetric (DSC)**

**Ion-moderated partition (*IMP*) chromatography**

**Size exclusion chromatography (SEC)**

**Pyrolysis-GC-MS**

**Liquid chromatography-mass spectrometry (LC-MS)**

**Scanning electron microscopy (SEM)**

**Confocal microscopy (CM)**

A decorative header featuring a warm, golden-yellow background. On the left, a large, stylized sun or moon is partially visible, with a soft, glowing aura. To the right, several stalks of wheat or grain are depicted, some standing upright and others falling, with individual grains scattered around them. The overall style is painterly and evokes a sense of harvest or natural abundance.

# **Products of Interest**



# Products of Interest

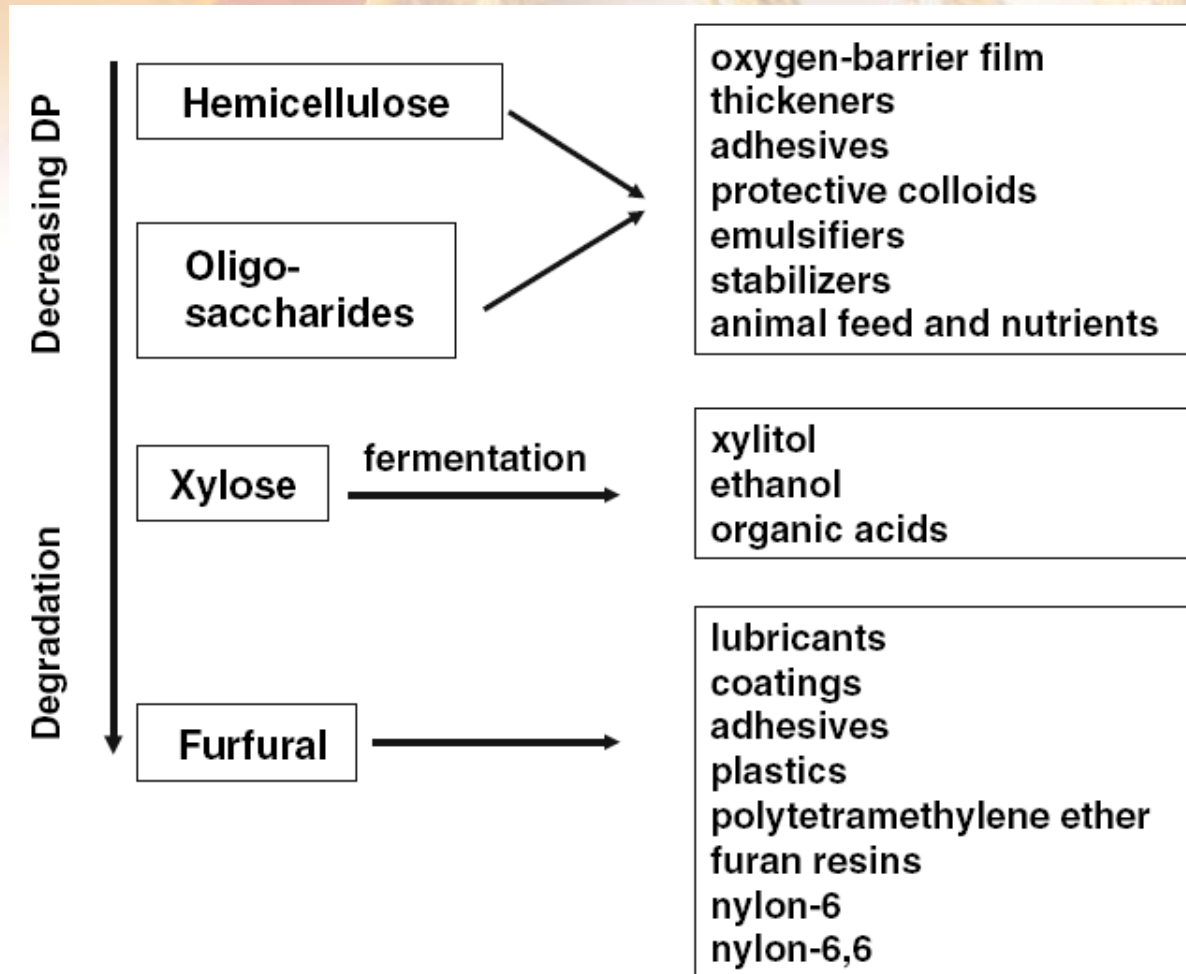
**Nutraceuticals**

**Functional Food Ingredients**

Cosmetics: Skin Care, Hair Care & Body Care

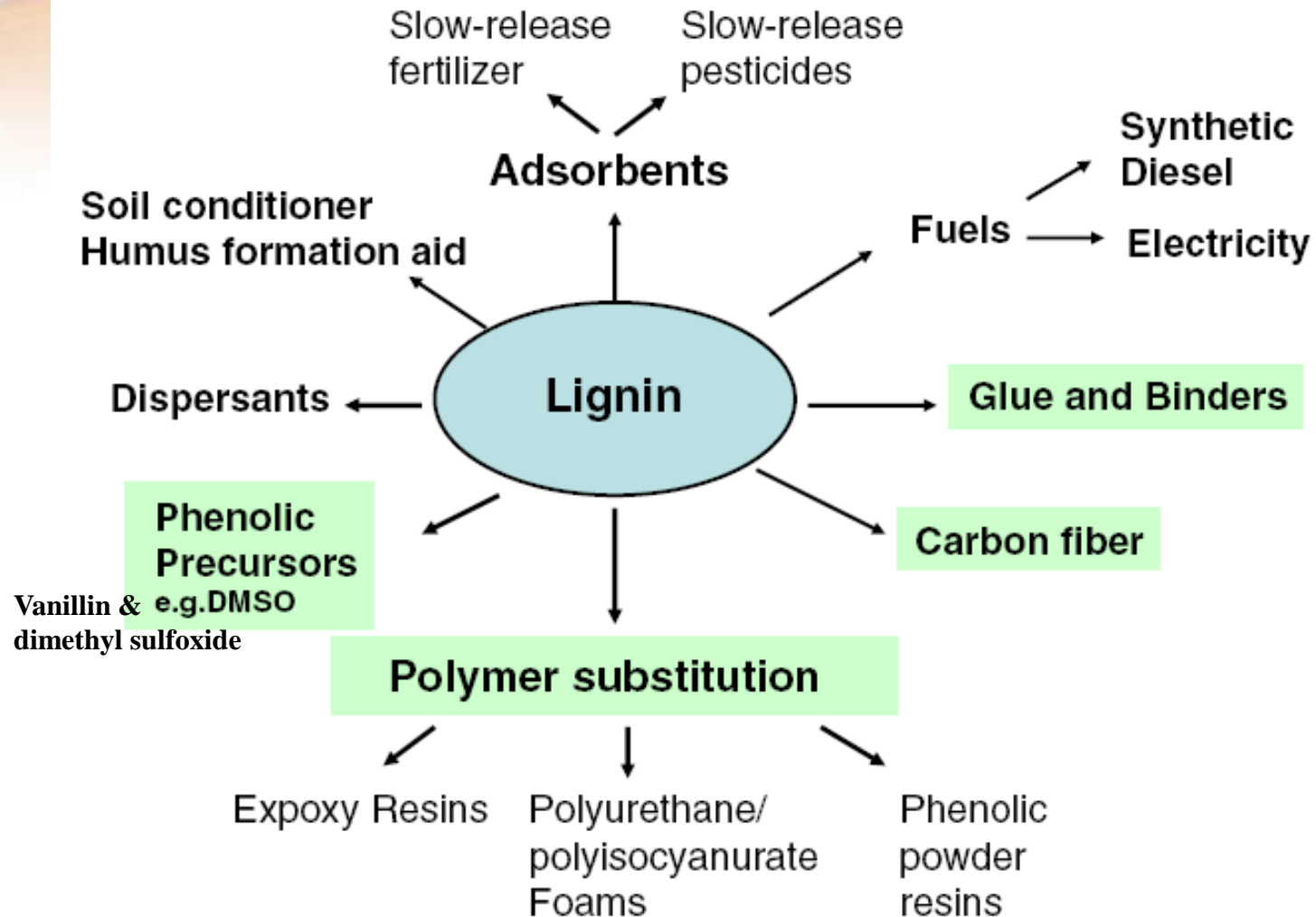
Medicinal, therapeutic & industrial products

# A hemicellulose utilization tree



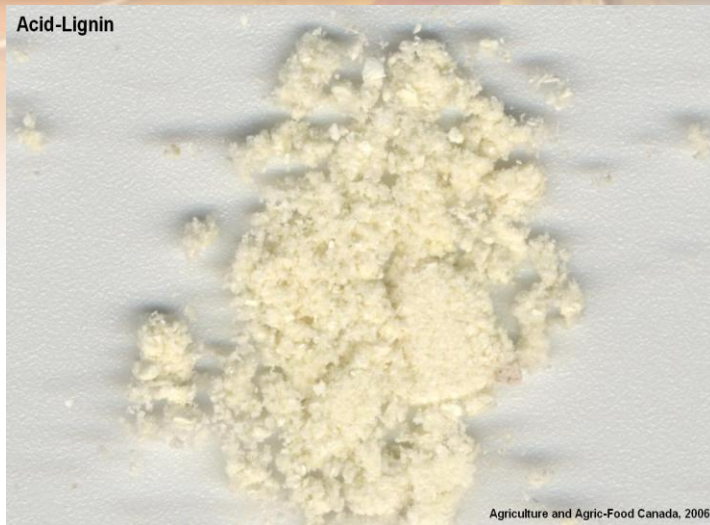


# A lignin utilization tree



# Biochemicals from Flax/Linseed Shives/Straw

Acid-Lignin



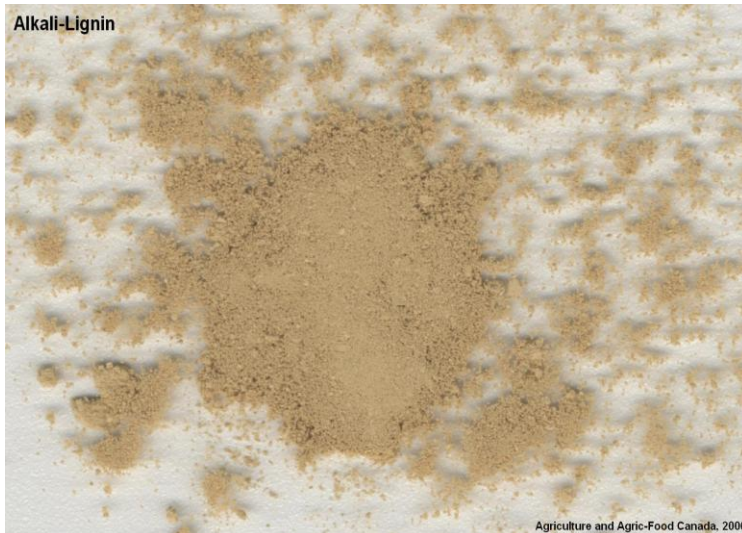
Agriculture and Agric-Food Canada, 2006

Hemicellulose



Agriculture and Agric-Food Canada, 2006

Alkali-Lignin



Agriculture and Agric-Food Canada, 2006

Cellulose



Agriculture and Agric-Food Canada, 2006

# Challenges

- **Developing sustainable green processes**
- **Chemical, physical and enzymatic modification**
- **Finding of specific functional/biological properties of components/molecules**

# Conclusions

**A variety of ‘green’ extractions and separation techniques are being investigated for manufacturing nutraceuticals, functional food ingredients, cosmetics, medicinal, therapeutic &/or industrial products**

**Selection and commercialization of the most appropriate technique(s) is being determined primarily by the raw material to be processed, phytochemical desired, and value of the finished product**



# Recent Scientific Publications

- Pronyk, C. and **G. Mazza**. 2009. Design and scale-up of pressurized fluid extractors for food and bioproducts. *Journal of Food Engineering*. Article in Press, Corrected Proof .  
[doi:10.1016/j.jfoodeng.2009.06.002](https://doi.org/10.1016/j.jfoodeng.2009.06.002)
- Kim, J-K.& **G. Mazza**. 2009. Extraction and separation of carbohydrates and phenolic compounds in flax shives with pH-controlled pressurized low polarity water. *Journal of Agricultural and Food Chemistry* 57, 1805 – 813.
- Athukorala, Y., **G. Mazza** & B. D. Oomah. 2009. Extraction, purification and characterization of wax from flax (*Linum usitatissimum*) straw. *European Journal of Lipid Science and Technology*, 111, 705–714.
- Buranov, A. U. & **G. Mazza**. 2009. Extraction and Purification of Ferulic Acid from Flax Shives, Wheat and Corn Bran by Alkaline Hydrolysis and Pressurized Solvents. *Food Chemistry*, [doi:10.1016/j.foodchem.2009.01.059](https://doi.org/10.1016/j.foodchem.2009.01.059)
- Hosseinian, F. S. and **G. Mazza**. 2009. Triticale bran and straw: Potential new sources of phenolic acids, proanthocyanidins, and lignans. *Journal of Functional Foods* 1, 57 –64.
- Güçlü-Üstündağ, Ö. and **G. Mazza**. 2009. Effects of Pressurized Low Polarity Water Extraction Parameters on Antioxidant Properties and Composition of Cow Cockle Seed Extracts. *Plant Foods Hum Nutr* 64:32–38.
- Karacabey, E. and **G. Mazza**. 2009. Optimization of antioxidant activity of grape cane extracts using response surface methodology. *Food Chem.*  
<http://dx.doi.org/10.1016/j.foodchem.2009.06.029>



# Recent Scientific Publications

- Karacabey, E. and **G. Mazza**. 2008. Optimization of Solid-Liquid Extraction of Resveratrol and Other Phenolic Compounds from Milled Grape Canes (*Vitis vinifera*). J. Agric. Food Chem. 56, 6318–6325.
- Buranov AU, Mazza G. 2008. Lignin in straw of herbaceous crops. Industrial Crops and Products 28, 237-259.
- Kim, J-K.& **G. Mazza**. 2008. Optimization of phosphoric acid catalyzed fractionation and enzymatic digestibility of flax shives. Industrial Crops & Products 28:346-355.
- Ho, C. H. L., Cacace, J. E. and **G. Mazza**. 2008. Mass transfer during pressurized low polarity water extraction of lignans from flaxseed meal. Journal of Food Engineering 89; 64–71.
- Rayne, S., Karacabey, E., and **G. Mazza**. 2008. Grape cane waste as a source of trans-resveratrol and trans-viniferin: High-value phytochemicals with medicinal and anti-phytopathogenic applications. Industrial Crops and Products 27 (3): 335-340.
- Rayne, S. and **G. Mazza**. 2008. Trichoderma reesei derived cellulase activity in three N,N-dimethylethanolammonium alkylcarboxylate ionic liquids. Nature Precedings.  
<http://precedings.nature.com/documents/632/version/1>.
- Mazza, G.** and T. Cottrell. 2008. Carotenoids and cyanogenic glucosides in saskatoon berries. Journal of Food Composition and Analysis 21:249–254.
- Güçlü-Üstündağ, Ö. and **G. Mazza**. 2008. Extraction of Saponins and Cyclopeptides from Cow Cockle Seed with Pressurized Low Polarity Water. LWT - Food Science and Technology 41 (9):1600-1606.

# Acknowledgements

## The research team



# **Acknowledgements**

## **The research team**

**Lana Fukumoto**

**David Godfrey**

**Anvar Buranov**

**Yasantha Athukorala**

**Farah Hosseinian**

**Carl Pronyk**

**Kelly Ross**

**Yukihiro Tamaki**

**Dongbao (David) Fu**

**Michael Parker**

**Miao Zhang**

**Robert Y. Nsimba**



# Pilot Plant South View





# Pilot Plant North View

